Critical Reactor Laboratory

Temperature Coefficient of Reactivity

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ABSTRACT

The temperature coefficient of reactivity of the RPI Critical Facility was measured in the range from 58.3 to 108 degrees Fahrenheit. The coefficient is small and positive below approximately 95°F and small and negative above that. It is presumed that the temperature coefficient remains negative thereafter.

PURPOSE:

A negative temperature coefficient is important to the safety of a reactor since it adds inherent stability with respect to temperature changes, such as might be induced in an excursion. The purpose of this experiment is the measurement of the temperature coefficient of reactivity of the RPI Critical Facility.

THEORY:

The reactivity of a reactor is defined as

The temperature coefficient of reactivity of, is the rate of change of reactivity with temperature.

Thus of the reactivity with temperature.

Since k is close to 1, of a pproximately of the rate of the rate of the reactivity with temperature.

The multiplication factor of a bare thermal reactor is $k = k P_T P_F = \eta_T f p \in P_T P_F P_T = 1/(1 + L_T^2 B^2)$ and

 $P_F = \exp(-B^2 \tau)$ are the thermal and fast non-leakage probabilities.

Thus $\ln k = \ln k_{\infty} + \ln P_{T} + \ln P_{F}$.

Differentiating term by term gives: $\alpha_{T} = k \frac{dR}{dT} = k_{\infty} \frac{dR}{dT} + \frac{dR}{R} \frac{dR}{dT} + \frac{dR}{R} \frac{dR}{dT}$ or $\alpha_{T} = \alpha_{T}(R_{\infty}) + \alpha_{T}(R_{F}) + \alpha_{T}(R_{F})$

Each of these temperature coefficients can be further differentiated so that $\alpha_T = \alpha_T(\eta_T) + \alpha_T(f) + \alpha_T(f) + \alpha_T(g) + \alpha_T(g) - \frac{B^2 L_T^2}{1 + B^2 L_T^2} \left[\alpha_T(L_T^2) + \alpha_T(B^2) \right] - B^2 T_T \left[\alpha_T \left(T_T \right) + \alpha_T(B^2) \right]$

For the bare thermal reactor, reactivity changes are explainable by using these coefficients. One can immediately guess that the temperature coefficients of p and will be unimportant due to the small concentration of U-238 in the fuel. This is the case and is also one of the causes of difficulty with the experiment. The resonance escape probability is one of the largest contributors to a negative temperature coefficient due to Doppler broadening of the resonances. In such small concentrations of U-238, (1) becomes less important relative to the other coefficients which may be positive or negative. For this reason the temperature coefficient of this reactor is very small, and may be of either sign in the temperature range investigated.

a) Temperature coefficient of η_{τ} $\eta_{\tau} = 0$ η_{τ} Nu is essentially constant at thermal energies so the temperature dependence of eta is the variation of the ratio η_{τ} with temperature. This is very small in the range in which we measured, as shown below.

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$$\frac{\sqrt{56}}{\sqrt{5}} \approx 0 \frac{94}{9a} = \frac{.9581}{.9610} = \frac{.9759}{.9780} \approx -10^{-5} / °C$$

b) Temperature effects on thermal utilization Lamarsh shows that $\alpha_{\tau}(f) = -(l-f)[\alpha_{\tau}(f) - \beta_{m}]$ where ζ is the thermal disadvantage factor and β_{m} is the coefficient of expansion of the moderator. $\alpha_{\tau}(\zeta)$ is always negative and $\beta_{H_{20}}$ is positive in the temperature range of interest so $\alpha_{\tau}(f)$ is always positive.

The temperature effects on the resonance escape probability

The temperature coefficient of p is always negative and
is a function primarily of the resonance integral and the coefficient of expansion of the moderator. As the temperature increases, moderator is expelled from the unit cell, effectively
increasing the fuel to moderator ratio. p decreases therefore
as the relative resonance absorber concentration increases.

d) Temperature effects on the fast fission factor Thermal expansion of the fuel increases slightly the probability that fast neutrons escape from the fuel. An increase in temperature also tends to flatten the thermal flux in the fuel, changing the spatial distribution of the primary fissions which decreases the probability that primary fission neutrons escape the fuel. These two competing factors combine to make $\forall_{\mathsf{T}}(\mathcal{E})$ small, and in this reactor, negligible.

e) Temperature effects on the non-leakage probabilties The buckling B² decreases with increasing temperature since the reactor dimensions increase. This means that neutron leakage decreases giving a positive $\langle \tau(\beta^2) \rangle$. However, the reactor structure tends to expand very little so that $\alpha_{\tau}(B^2)$ is generally very small. Since $\alpha_{\tau}(L_{\tau}^2)$ and $\alpha_{\tau}(\tau_{\tau})$ are both positive, the net temperature coefficients of the nonleakage probabilties are negative.

PROCEDURE:

A three rod bank was raised to approximately 20 inches and the reactor was brought citical on the remaining rod which was used thereafter to relate all reactivity changes. The two 18 kw heaters were turned on (the agitators had been on since the beginning to maintain an even temperature and to avoid perturbing the core as they were turned on). After the response of the reactor to temperature was noted (positive or negative period) negative or positive reactivity was added with the control rod to put the reactor on a small but opposite period. Due to the effect of the temperature coefficient, the reactivity added is eventually balanced as the reactor passes through critical. The temperature coefficient of reactivity between the two successive critical positions and temperatures is:

 $\frac{\partial f}{\partial T} = \left[\frac{(4/\text{in.})_f + (4/\text{in})_i}{Z} \right] \left[\frac{(\text{inches})_i - (\text{inches})_f}{T_f} \right]$

This procedure is followed throughout the temperature range by relating successive critical positions and the corresponding temperatures. The coefficient applies at the midpoint of the initial and final temperatures.

DISCUSSION:

One of the experimental difficulties encountered involved the amount of reactivity to be inserted, which the temperature coefficient had to overcome. If too much recativity is inserted a very long time would be required to return to critical at the slow rate at which the bulk temperature of the water increased. On the other hand, too little reactivity would introduce possible large numerical errors since numbers which are almost equal must be subtracted. This allows targe percentage errors to occur although the absolute error may be small. Based on trial and error, I think a one-half cent reactivity addition would be about the best compromise of these requirements.

The temperature coefficient of reactivity is one of the basic parameters affecting the stability of a reactor and is important both from a safety and from an operating standpoint. If the coefficient were positive, a power change would initiate reactivity additions causing the chain reaction to diverge. At best this would be a nuisance since the reactor would resist attempts to bring it to critical. At worst it would be a safety hazard which would allow the reactor to runaway Although a large negative temperature coefficient presents no safety problem, the possibility of power overshoots during power changes occurs. In this case, the temperature increase causes such a large negative reactivity change that the reactor falls subcritical and the power level drops until the reactivity is zero. If the coefficient is negative and small, the reactor is stable.

An important distinction should be made between the temperature coefficient measured in this experiment, and the one actually responsible for safety. Due to the slow rate at which the temperature of the water changes, and the fact that the fuel is contained in thin plates, all temperature changes can be considered uniform. During a power change, the fuel temperature responds almost instantaneously. It is the prompt temperature coefficient of the fuel which is responsible for safety, since

the moderator temperature changes more slowly.

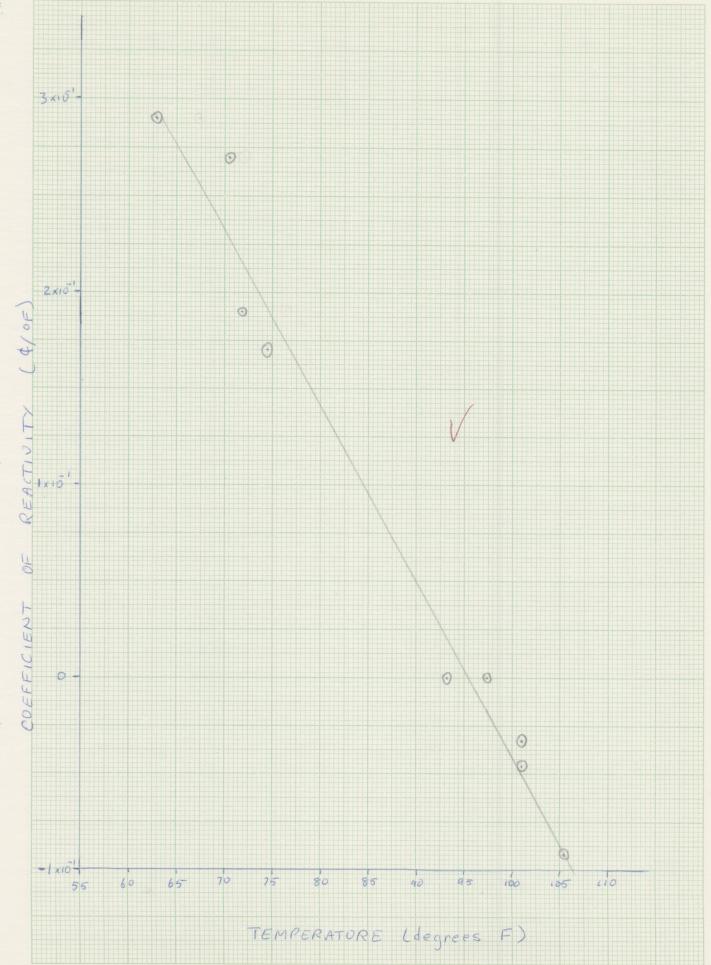
At lower temperatures the temperature coefficient is positive, is zero between 90 - 100 degrees F, and is therafter negative. As stated, the temperature coefficient of f is always positive. Since the coefficients of 1, E, p, are negligible, x_(Ko) is positive and must be compensated for by leakage. Although the temperature coefficients of the following they do not equal that of f at first, possibly because they involve a factor of B², which is a small number. As the temperature increases further leakage increases and dominates due to the fact that at lower moderator density the mean free path for scattering increases allowing more neutrons to escape. Although to have an infinite reflector due though the temperature coefficients of the leakage are negative, to the large amount of water surrounding the core, reflectors of other reactors would become less effective if they had a fixed volume since some of the reflector would be expelled. At much higher temperatures the temperature coefficient would presumably remain negative. It is difficult to extrapolate the prompt temperature coefficient from the data, but it should also be negative since the most important factors affecting the fuel, eta andp, have negative coefficients. A power excursion would then

then be controlled by the temperature coefficient, eventually or not of compensating for excess reactivity and inducing criticality, well be although at a possibly high power level. This of course assumes that no meltdown or other core disruptive accident occurs.

There is really no trade-off between safety and economecs as far as temperature coefficients are concerned. Although fuel loadings could be reduced slightly with a positive coefficient, and the core would be less reactive when clean and cold, the negative coefficient is needed at operating temperatures. In fact, when one considers the effort to reduce the possibility of core meltdown, one of the worst credible accidents, designing a negative coefficient seems one of the cheap-

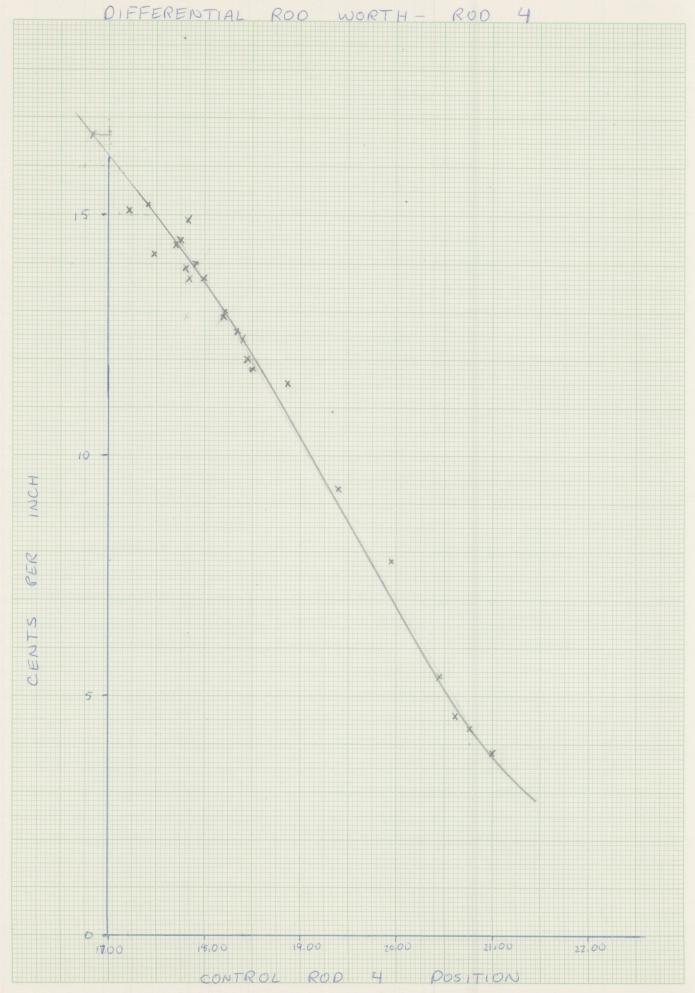
est ways of decreasing that probability.

temperature coefficient and for solid moderators this is through the thermal disadvantage factor $\int_{-2}^{2} \sqrt{\Phi_{p}}$. If $\int_{-2}^{2} decreases$ with temperature, $\nabla_{-1}(f)$ becomes more positive. As stated previously, this does occur since the flux tends to flatten as the diffusion length increases with temperature. If you assume that the moderating properties of liquid water acting only as a coolant do not change much, the contributions to the coefficient from leakage would depend only on the properties of the solid moderator. These might not change as rapidly with T as with a liquid moderator which has larger density changes. The net result could be a positive temperature coefficient.



4 - 5	+2.9×10	1-01x p.1+	+ 2.7 × 10-1	+1.7 × 10-1	0	2-01×9°17-	0	- 3.3×10-2	-4.3×10
FINAL CRITICAL POSITION	16.790	16.687	19.862	19.753	19.969	20.043	909.81	18.615	18.659
INITIAL CRITICAL POSITION	16.951	16.790	186.61	19.362	696.61	19.969	909'81	18:606	18:615
FINAL JEMP.	67.4	76.41	72.1	0.17	95.5	106.6	99.3	102.5	108:0
TEMPERATURE (°F)	58.3	4.60	0.69	72.1	9101	95.5	95.0	99.3	102.5
BANK POSITION (In)	2 2.00	22.00	19.25	19.25	19.15	19.15	19.65	19.65	19.65

TEMPERATURE COEFFICIENT DATA



Critical	SuperCritical				
Position	Position	Midpoint	Period	4	4/in
17.76	19.00	18.38	55	15.4	12.4
16.90	17.51	17.21	109	9.2	15.1
17.145	17.805	17.475	105	9.4	14.2
17.392	18.005	17.699	112	8.8	14.4
17.392	19.295	18.344	27	24	12.6
17.36	18.32	17.84	61.0	14.3	14.9
16.89	18.68	17.79	25.4	24.8	13.9
17.39	18.28	17.84	75	12.2	13.7
18.015	19.705	18.86	38.6	19.4	11.5
16.840	18.000	17.42	45	17.6	15.2
16.95	18.50	17.73	30	22.5	1405
17.20	18.80	18.00	31.3	21.9	13.7
17.40	19.00	18.20	35	20.6	12.9
17.49	19.50	18.5	27.8	23.6	11.8
17.40	19.50	18.45	24.7	25.3	12.0
17.75	18.50	17.88	46.0	17.5	14.0
19.280	21.960	20.62	70	12.4	4.6
20.100	21.945	21.023	150	7.05	3.8
19.570	21.960	20.77	94	10.25	4.3
18.930	21.960	20.445	51	16.4	5.4
17.25	19.00	18.22	29.4	27.7	13.0
18.80	20.00	19.40	83.5	11.2	9.3
18.87	21.00	19.94	48.4	16.6	7.8
ROD 4.	Calibrat	tion Oata			

CALCULATION: Heating Rate of Water: Neglecting structure, losses to atm Total Heat Rate = 36 Kw = map of = pVapot/oo assume p, ap independent of tempurature et 36 km @ 20°C, p= 1000.52 kg cp= 4.1818×103 W-s V = 2000 gallons x 231 in3 1 m3 = 7.57 m3 at = 36×103 W a0 = 103 26 × 7.57 × 4.18×103 W-s = 1.14×10 3 oK = 6.87×102 °K = 4.09 °K = 7.37 65